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A. The Number Of Reactant Collisions Per Unit Time B. The Energy Of Each Collision C. The Orientation Of The Particles In Each Collision D. The Temperature E. All Of The Above 10. Which One Of The Following Statements Is True? A. Competitive Inhibition Is Reversible And Can Be 4th, 2024

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May Be More Efficient Than Another, Regardless Of The Thermodynamics Of The

Reaction, So You Do Not Have Enough Information To Answer The Question. D) Assume That You Have A Reaction $X \rightarrow Y$ Where $\Delta G = 0$ And The Energy Of Activation = $+0.6$ Kcal/mole. In A Cell, You Find That The React 3th, 2024

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