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 C_p, C_v, P, V, T ? As A Sample Application, We Can Assume The Ideal Gas Equation Of State. $V = RT/P \Rightarrow (\partial V / \partial T)_P = R/P$. Substitution $\Rightarrow \Delta S = C_p \ln(T_2/T_1) - R \ln(P_2/P_1)$ Applying
 $\Delta S = 0 \Rightarrow \ln(T_2/T_1) = R/C_p \ln(P_2/P_1) \Rightarrow (T_2/T_1) = (P_2/P_1)^{C_p/C_p - C_v} = (P_2/P_1)^{C_p/C_p - C_v}$ 13th, 2024.

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