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In Which Any Number Can Be Represented In The Form  $N \times 10^N$ . N (Where N Is An Exponent Having Positive Or Negative Values And N Can Vary Between 1 To 10). E.g. We Can Write 232.508 As  $2.32508 \times 10^2$ . In Scientific Notation. Similarly, 0.00016 Can Be Written As  $1.6 \times 10^{-4}$ . Precision Refers 13th, 2024

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Class–XI–CBSE–Chemistry Thermodynamics CBSE NCERT Solutions For Class 11 Chemistry Chapter 6 Back Of Chapter Questions 1. Choose The Correct Answer. A Thermodynamic State Function Is A Quantity (i) Used To Determine Heat Changes (ii) Whose Value Is Independent Of Path (iii) Used To Determine Pressure-volume Work 16th, 2024

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Class- XII-CBSE-Chemistry Haloalkanes And Haloarenes Practice More On Haloalkanes And Haloarenes Page - 12 Www.embibe.com Page No.310 1. Name The Following Halides According To IUPAC System And Classify Them As Alkyl, Allyl, Benzyl (primary, Secondary, Tertiary), Vinyl Or Aryl Ha 10th, 2024

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CBSE NCERT Solutions For Class 11 Chemistry Chapter 4 Back Of Chapter Questions 1. Explain The Formation Of A Chemical Bond. Solution: A Chemical Bond Is Defined As An Attractive Force That Holds The Constit 5th, 2024

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Arrange The Following Polymers In Increasing Order Of Their Intermolecular Forces (i) Nylon 6, 6, Buna-S, Polythene (ii) Nylon 6, Neoprene, Polyvinyl Chloride. Solution: On The Basis Of Intermolecular Forces Of Attraction, The Polymers Can Be Classified Into Elastomers, F 5th, 2024

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Mendeleev Arranged The Elements In His Periodic Table According To Their Atomic Weight Or Mass. He Arranged The Elements In Periods And Groups In Order Of Their Increasing Atomic Weight. He Also Kept The Elements With Similar Properties In The Same Group. But, He Found Out That If The Elements Were 7th, 2024

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$=E(H+H/2)0 - RT NF \ln H_2 [H+]$  Class- XII-CBSE-Chemistry Electrochemistry Practice More On Electrochemistry Page - 3 Www.embibe.com ... Has Ecell  $0 = 0.236 \text{ V}$  At 298 K. Calculate The Standard Gibbs Energy And The Equilibrium Constant O 5th, 2024

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Starch Is A Carbohydrate Consisting Of Two Components - Amylose (15 - 20%) And Amylopectin (80 - 85%) Whereas Glycogen Consists Of Only One Component Whose Structure Is Similar To Amylopectin. 14.7. What Are The Hydro 2th, 2024

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