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Chemical Reaction • A Chemical Reaction Always Involves A Chemical Change Because Atoms Of The Reacting Substances Form New Combinations With New Properties. • For Example, A Chemical Reaction (and Chemical Change) Takes Place When A Piece Of Iron (Fe) Combine 3th, 2024

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### **Chemical Quantities**

1-carbon 2-oxygen Thus One Mole Of CO<sub>2</sub> Contains Three Times Avogadro's # Of Atoms. Example: How Many Atoms Are In 2.12 Mole Of Propane (C<sub>3</sub>H<sub>8</sub>)? Know: # Of Moles = 2.12 Mole C<sub>3</sub>H<sub>8</sub> 1 Mole C<sub>3</sub>H<sub>8</sub> = 6.02 X 10<sup>23</sup> Molecules C<sub>3</sub>H<sub>8</sub> 1 Molecule C<sub>3</sub>H<sub>8</sub> = 11 Atoms ( 3 Carbon Atoms & 8 Hydrogen Atoms) 2th, 2024

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