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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 Cm<sup>3</sup> At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And Mar 1th, 2024

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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established.  $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$  What Would Alter The

Composition Of The Equilibrium Mixture In Favour Of The  $V^{2+}$  Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Jan 1th, 2024

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Fahad H. Ahmad (whatsapp: +92 323 509 4443)  
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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s).  
First Ionisation Energy Of Silver =  $+731\text{kJmol}^{-1}$  Electron Affinity Of Bromine =  $325\text{kJmol}^{-1}$  Enthalpy Change Of Atomisation Of Silver =  $+285\text{kJmol}^{-1}$  Enthalpy Change

Of Atomisation Of Bromine =  $+112\text{kJmol}^{-1}$  ... Mar 1th, 2024

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