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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As

The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 Cm³ At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And Mar 11th, 2024.

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CHEMISTRY 9701/01 1 Hour10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Apr 25th, 2024.

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Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following
Data To Calculate A Value For The Lattice Energy Of AgBr(s). First Ionisation Energy
Of Silver = +731 kJ mol⁻¹ Electron Affinity Of Bromine = 325 kJ mol⁻¹ Enthalpy Change
Of Atomisation Of Silver = +285 kJ mol⁻¹ Enthalpy Change Of Atomisation Of Bromine
= +112 kJ mol⁻¹ ... Jan 9th, 2024.

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