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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 cm³ At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And Mar 2th, 2024

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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of

The V²⁺ Ions? A Adding An Acid B Adding A Reagent
That Selectively Precipitate Mar 2th, 2024

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(c) (i) Write A Chemical Equation Representing The

Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of Atomisation Of Bromine = +112kJmol⁻¹ ... Jun 1th, 2024

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