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black colour should be seen as the starch reacts with the residual iodine. Continue to add FA 1 (1 cm³) at a time until the blue-black colour of the starch-iodine complex disappears. Feb 1st, 2024
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When vanadium(II) compounds are dissolved in water, the following equilibrium is established.
$$\text{V}^{2+} + \text{H}_2\text{O} \rightleftharpoons \text{V}^{3+} + \text{H}^+$$

What would alter the composition of the equilibrium mixture in favour of the V^{2+} ions?
A Adding an acid
B Adding a reagent that selectively precipitates
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Chemical Equation Representing The Lattice Energy Of
AgBr. (ii) Use The Following Data To Calculate A Value
For The Lattice Energy Of AgBr(s). First Ionisation
Energy Of Silver = +731 kJ mol⁻¹ Electron Affinity Of
Bromine = 325 kJ mol⁻¹ Enthalpy Change Of
Atomisation Of Silver = +285 kJ mol⁻¹ Enthalpy Change
Of Atomisation Of Bromine = +112 kJ mol⁻¹ ... Apr 1th,
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