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12 The Gas Laws Name Period Date THE IDEAL GAS LAW N PV = NRT Where Pressure In Atmosphere Volume In Liters = Number Of Moles Of Gas Universal Gas Constant = 0.0821 Atm/mol.K Kelvin Tem Erature 1. 2. U 150 5. 7. 8. 9. How M Will Occupy A Volume Of 2.50 Liters At 1.20 Atm And 25 0 C? Moles Of 4th, 2024

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2 In The Strings Must Satisfy Which Of The Following Relations? (A) T 1 = T 2 (B) T 1 >T 2 (C) T 1