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2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{Aniline}) = n(\text{Acetanilide}) = 0.0322$.
Ide Mol We Find The Molar Mass Of Acetanilide: $(12 \times 8) + (1 \times 9) + (14 \times 1) + (16 \times 1) + 135 = 177$.
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