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Chemistry 11 Stoichiometry Review Worksheet Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \text{ O}_2 \times 1 \text{ mol } 32.00 \text{ g} \times 1 \text{ mol} = 3.11 \text{ g O}_2$ %Yield = $\frac{1.45 \text{ g } 3.11 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%? pure FeCO_3 Jan 1th, 2024

Basic Stoichiometry Phet Lab Answer Key

Basic Stoichiometry Phet Lab Answer Key Page 15/28 Read Online Basic Stoichiometry Phet Lab Answer Key (ANSWER 386.3g Of LiNO_3) 4) Using The Following Equation: $\text{Fe}_2\text{O}_3 + 3 \text{H}_2 \rightarrow 2 \text{Fe} + 3 \text{H}_2\text{O}$. Calculate How Many Grams

Of Iron Can Be Made From 16.5 Grams Of Fe_2O_3 By The Following Equation. May 4th, 2024

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