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Stoichiometry

Mass Percentage Of Carbon In Ethanol? Chapter 4
Molar Mass = $2 \times 12.011 + 6 \times 1.008 + 1 \times 15.999 = 46.069$ G/mol
% C = $2 \times 12.011 \times 100 / 46.069 = 52.14$ %
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