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 $3) 2(aq) + Y C. K(s) + H_2SO_4(aq) + Y D. Zn(s) + NaOH(aq) + Y$
SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION
1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate, $Ba(NO_3)_2$, And Sodium Sulfate, Na_2SO_4 .
2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction.
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6.1 - Describing Chemical Reactions (b) This Is A Chemical Reaction As A Gas Is Created (carbon Dioxide) During The Reaction. (a) + H₂O (g) (b) The Bottle Could Be Heated Up Or Opened To Reverse The Reaction. CHECK YOUR LEARNING Suggested Answers 1. The Arrow Shows Which Way The Chemical

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