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### **Electron Arrangement Of The Electron Configuration Of Atoms**

The Distance From The Bottom Of The Chart Indicates The Energy Of Each Energy Level And Sub-level. The Closer The Energy Level Is To The Bottom Of The Chart, The Lower Its Energy. At The Bottom Of The Chart, You Will find The first Energy Level,  $N = 1$ . The Chart Shows Only One Circle In The first Energy Level. 1th, 2024

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1 S22s 2p63 23p64s1 C. Gallium 1s 22s 22p63s 3p63d104s 4p1 D. Copper 1s 22s 2p 63s23p 3d104s1 PROBLEMS Write The Answer On The Line To The Left. Show All Your Work In The Space Provided. 9. 1 1012 M What Is The Wavelength Of Light That Has A Frequency Of 3 10 4 Hz In A Vacuum? 10. 3.3 10 19 3th, 2024

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