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Excess Of Ethanol Was Used Because Its Lower Critical

Point Reduces The Reaction Middle, Making This Reaction Technically Viable. The Estimates Of The Critical Temperature And Pressure Of The SODD And Ethanol Mixture Were Presented In Section 2.1 And The Values Are Shown In Table 2. At 553.15K, The Molar Ratios Of 10, 20 And 30 Lead Jun 1th, 2024

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In Your Reaction, The Synthesis Of Banana Oil - Isopentyl Acetate, You Will Be Using Isopentyl Alcohol (3-methyl-1-butanol) And Acetic Acid (ethanoic Acid, See Figure 3). $\text{H}_3\text{C}-\text{O}-\text{O}-\text{H} + \text{HO}(\text{CH}_2)\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3 + \text{HO}-\text{H}$ Acetic Acid (ethanoic Acid) MW 60.1, D 1.05 Bp 118 O C Isopentyl Alcohol (3-methyl-1-butanol) MW 88.1, D 0 ... Feb 1th, 2024

22. The Fischer Esterification

In This Laboratory Experiment, You Will Prepare Banana Oil (isopentyl Acetate) By The Fischer Esterification Of Acetic Acid And Isopentyl Alcohol (figure 3). The Reaction Will Be Catalyzed By The Addition Of Sulfuric Acid. Once The Reaction Is Complete, A Distillation Will Be Performed To ... Apr 1th, 2024

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$W_{Eq} = E W E W C A B A B W J V S K C C K C C C D t D C 0$
1 – = – $\gamma \gamma \gamma \gamma$ (6) The Relationship Between Flux And Feed Water Concentration Is Nonlinear And Varies Throughout The Reaction. Hence The Following Modified Relation Can Be Jun 1st, 2024

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