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23 0.0435 12 0.174 16 0.00726 0.003625 0.003625 0.003625 0.01088 0.003625  
Empirical Formula =  $\text{Na}_2\text{CO}_3$  2 1 3 5) 0.150 G Of Copper Reacts With Oxygen Form 0.188 G Of Copper Oxide. Find The Empirical Formula Of Copper Oxide. Mass Cu = 0.150 G Cu O Mass O = 0.188 - 0.150 = 0.038 G 0.150 63.5 0.038 16 0.00236 0.00236 0.002375 0.00236 Mar 4th, 2024.

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