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Ideal Gas Law Worksheet $PV = nRT$ Gas Laws Packet Ideal Gas Law Worksheet $PV = nRT$. Use The Ideal Gas Law, " $PV = nRT$ ", And The Universal Gas Constant $R = 0.0821 \text{ L}\cdot\text{atm} / (\text{K}\cdot\text{mol})$. To Solve The Following Problems: K = mol. If Pressure Is Needed In kPa Then Convert By Multiplying By $101.3 \text{ kPa} / 1 \text{ atm}$ to Get. $R = 8.31 \text{ L}\cdot\text{kPa} / (\text{K}\cdot\text{mole})$ | Mar 2th, 2024

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