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Metal ... IONIC BONDING Name Ionic Bonding Occurs
When A Metal Transfers One Or More Electrons To A
Nonmetal In An Effort To Attain A Stable Octet Of
Electrons. For Example, The Transfer Of An Electron
From Sodium To Chlorine Can Be Depicted By A Lewis
Dot Diagram, Calcium Would Need Two Chlorine Atoms
To Get Rid Of Its Two Valence Electrons, Jan 8th,
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Transfer Of Valence Electron To Attain Stability. This
Type Of Bonding Leads To The Formation Of Two
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Bonding = Ca 3 P 2 Formula Unit This Is A Chemical Formula That Shows The Types And Numbers Of Atoms

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