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The Number Of Recorded Counts. This Variation Is Called The Statistical Uncertainty Of A Measurement. It Is An Uncertainty That Is Well Understood. Consider A Measurement That Is Done For A Time Period  $\Delta t$ . Suppose That There Are  $N$  Radioactive Nuclei, And That Each One Has A Probability  $P$  To Decay In The

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Think Of As Hydrogen, Is Called Hydrogen-1 And Is Stable. The Other Two Isotopes Are Called Hydrogen-2 (deuterium) And Hydrogen-3 (tritium). Deuterium Feb 1th, 2024

## **Introduction To Nuclear Physics And Nuclear Decay**

The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In Mind That ... (two Protons And Two Neutrons) ! " $= 2 \text{ } ^4\text{He} + 2 \dots$  Or An Anti-neutrino ( ) Neutrinos Have No Charge, Spin  $1/2$ , And Mass  $\sim 0.1 - 1 \text{ eV}$  (?) The Beta Particle Feb 1th, 2024

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## **CHAPTER 1 Radioactive Decay & Decay Modes**

This Equation Represents A First-order Reaction And In  
The Present Case A Mononuclear Reaction. Radioactive  
Decay Is Only Possible If  $\Delta E > 0$ .  $\Delta E$  Can Be  
Determined By Comparison Of The Masses. According  
To The Relation By Einstein ( $E = Mc^2$ ): (EQ 3) By  
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$N(t) = N_0 e^{-kt}$ ;  $\ln \frac{N(t)}{N_0} = -kt$  Where  $N(t)$  Is The Number Of Atoms Remaining At Time  $t$ ,  $N_0 = N(0)$  Is The Initial Number Of Atoms, And  $k = \frac{\ln(2)}{T_{1/2}}$  Is A Constant Determined By The Nuclear Half-life  $T_{1/2}$ . However, Recent

Work Has Uncovered Evidence Of An Unexpected Periodicity In Several Nuclear Decays [2 Jan 7th, 2024

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Nuclear Decay Organizer Students Know The Three Most Common Forms Of Radioactive Decay (alpha, Beta, And Gamma) And Know How The Nucleus Changes In Each Type Of Decay. Students Know Alpha, Beta, And Gamma Radiation Produce Different



Amounts And Kinds Of Damage In Matter And Have Different Penetrations. Students Know Some Naturally Occurring Isotopes Of Elements Are Radioactive, As Are ... May 18th, 2024

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Balancing Decay Reactions: Fill In The Blank In Each Of The Following Decay Reactions With The Correct Decay Particle Or Decayed Nucleus That Will Balance The Reaction, Also State Whether It Is Alpha Or Beta Decay

15.  ${}^{184}_{74}\text{W}$   $\rightarrow$   ${}^{180}_{72}\text{He}$  + \_\_\_\_\_ Is An Example Of \_\_\_\_\_ Decay.

16.  ${}^{35}_{16}\text{S}$  + \_\_\_\_\_  $\rightarrow$   ${}^{35}_{17}\text{Cl}$

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