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1

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• Arranged Elements By Increasing Atomic Mass • Noticed The Repetition Of Properties Every Eighth Element • Created The Law Of Octaves Lothar Meyer (1830–1895) • Demonstrated A Connection Between Atomic Mass And Elements' Properties • Arranged The Elements In Order Of Increasing Atomic Mass Dmitri Mendeleev (1834–1907) Mar 9th, 2024

The Periodic Table: Periodic Trends

B Trends In The Periodic Table In Each Group, The Elements Have Similar Properties, With A Gradual Trend In Properties Down The Group. This Is Because They Have The Same Number Of Electrons In The Outer Shells. Elements On The Left Side Of The Periodic Table Are Metals And Those On The Right Side Are Non-metals. Feb 11th, 2024

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Periodic Trends Are Specific Patterns In The Properties Of Chemical Elements That Are Revealed In The Periodic Table Of Elements. Major Periodic Trends Include Electronegativity, Ionization Energy, Electron Affinity, Atomic Radii, Ionic Radius, Metallic Character, And Chemical Reactivity. May 12th, 2024

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