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A 1.5 Atm B. 3 Atm C. 6 Atm D. 0.547 Atm E. 0.333 Atm  $P_1V_1/T_1=P_2V_2/T_2$  Solve For Pressure P2 By Setting The Two Equations Equal To Each Other. Multiply P1 By V1 And Divide By Temperature T1. Then Do The Same Thing To P2, V2, And T2.:  $3 \text{ Atm}(1\text{m}^3) = P_2(2\text{m}^3)$   $100 \text{ K} = 400\text{K}$   $P_2 = 6 \text{ Atm}$  24th, 2024

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$P_1 V_1 = P_2 V_2$   $\rho_1 V_1 = \rho_2 V_2$ . The Pressure Difference  $\Delta P = \rho g h$  is Given In The Text As Well As The Density And Water Speed. Note That The Flow Speed Should Increase In The Narrow End Because  $A_1 v_1 = A_2 v_2$  ( ). This Observation Helps Us To Avoid Possible Mistakes With 12th, 2024

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