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WORKSHEET # 2: THE MOLE AS A UNIT OF MASS Define The Term Molar Mass
(worksheet #1): ____ Now That You Know How To Find The Mass Of One Mole Of A
Substance (molar Mass) You Can Easily Find The Mass Of Several Moles Or The Mass
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Information In This Topic, The Notes Is Going To Help You. A) Find The Mules Of The
Compound With Known Jan 4th, 2024Mole To Mole Stoichiometry Worksheet
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Chemical Reactions: A. $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$ B. $2 \text{KNO}_3 \rightarrow 2 \text{KNO}_2 + \text{O}_2$ C. $2 \text{O}_3 \rightarrow 3 \text{O}_2$ D.
 $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + 2 \text{H}_2\text{O}$ E. $4 \text{CH}_3\text{NH}_2 + 9 \text{O}_2 \rightarrow 4 \text{CO}_2 + 10 \text{H}_2\text{O} + 2 \text{N}_2$ F. $\text{Cr}(\text{OH})_3 + 3$
 $\text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + 3 \text{H}_2\text{O}$ W Feb 5th, 2024.

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Calculations From Chemical Equations Mole - Mole ... $7 + 6 \text{KI} + 7 \text{H}_2\text{SO}_4 \rightarrow \text{Cr}_2(\text{SO}_4)_3 + 4 \text{K}_2\text{SO}_4 + 3 \text{I}_2 + 7 \text{H}_2\text{O}$ A) How Many Moles Of Potassium Dichromate ($\text{K}_2\text{Cr}_2\text{O}_7$) Are Required ... = 407.9 G AgBr This Is The Theoretical Yield Yields 22 B)Calculate The Percent Yield If 375.0 G Of Silver Bromide Was Obtained From The Reaction Theoretical Yield = 407.9 G AgBr Percent Yield = $100 \times \frac{\text{Actual Yield}}{\text{Theoretical Yield}}$ Apr 1th, 2024

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