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Reactions Occur When The Elements In A Chemical Reaction Gain Or Lose Electrons,  
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Oxidation Reduction Examples With AnswersA.  $\text{ClO}_3^-$   $\text{ClO}_2^-$  Reductive Agent B.  
 $\text{SO}_4^{2-}$   $\text{S}^{2-}$  Reducting Agent C.  $\text{Mn}^{2+}$   $\text{MnO}_2$  Oxidizing Agent D.  $\text{Zn}^{2+}$   $\text{Cl}_2$  Oxidant Write  
Balanced Equations For The Following Redox Reactions: A.  $2 \text{NaBr}$  And  $\text{Cl}_2$   $\text{NaCl}$  And  
 $\text{Br}_2$  B.  $\text{Fe}_2\text{O}_3$  No  $3 \text{CO}$   $2 \text{Fe}$  No  $3 \text{CO}_2$  In Sour Solution C.  $5 \text{CO}$  And  $\text{I}_2\text{O}_5$   $5 \text{CO}_2$  And  
 $\text{I}_2$  In The Basi 7th, 2024Pyrite Oxidation And Its Control Amd Molecular Oxidation  
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Occur In Four Isomers:  $\alpha$ ,  $\beta$ ,  $\gamma$ , And  $\delta$ . The Amount And Distribution Of These Four  
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