

# Relative Mass And The Mole Answer Key Pdf Download

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Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) 1.  $N_2 + 2O_2 \rightarrow N_2O_4$  A. If 15.0g Of  $N_2O_4$  Was Produced, How Many Moles Of  $O_2$  Were Required?  $15.0g N_2O_4 \times \frac{1 \text{ Mol } N_2O_4}{2 \text{ Mol } O_2} = 0.326 \text{ Mol } O_2$  B. If  $4.0 \times 10^{-3}$  Moles Of Oxygen Reacted, How Many Grams Of  $N_2$

Were Needed?  $4.0 \times 10^{-3}$  Mol O 2 1 Mol N 2 28 ... Feb 2th, 2024.

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