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$2\text{Cl}_2 \rightarrow 2\text{HCl}$. 7 Chemistry-I. Figure 1.1: Plot Of Concentration Against T For A Reaction $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$ The Rate Of This Reaction Does Not Depend On Concentration. 1.6 First Order Kinetics A Reaction Of The first Order Is Represented As $X \rightarrow Y$ Where X Is The Reactant And Y The Product. The Rate Of The Reaction Will Be Directly Jan 16th, 2024

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