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4. HCl(aq) + Sr(OH)2(aq) ¬ SrCl2(aq) + H2O(I) 5.
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Polysaccharide Starch, Cellulose, Glycogen (animal) Benedicts-glucose Iodine-starch Amino Acid Protein, Polypeptide Hair, Enzyme, Hemoglobin, Insulin Ninhydrin, Biuret Nucleotide Nucleic Acid DNA, RNA Methyl Green Fatty Acid/glycerol Fat Or Tri - Glyceride Cooking Oil, Butter Grease Test With Brown Paper Feb 22th, 2024

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Over A Bunsen Burner Strongly For About 3 Minutes. Insert A Burning Wood Splint Into The Test Tube. If Carbon Dioxide Gas (CO 2) Is Present, It Will Put The Flame Out. Note Any Change In The Appearance Of The Residue In The Test Tube. DISPOSAL: This May

Safely Go Down The D May 7th, 2024

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3. Heat The Crucible With A Bunsen Burner Until No Further Change Is Observed. 4. Save The Contents For Reaction 4. Reaction 4 1. Allow The Crucible From Reaction 3 To Cool For 5-10 Minutes (until The Crucible Is Cool To The Touch). 2. Use A Medicine Jan 8th, 2024

LAB: TYPES OF CHEMICAL REACTIONS

E. Turn On The Methane Gas And Light The Bunsen Burner. Open The Air Holes To Add More Oxygen And Make A Blue Cone. (Use This Flame For The Next Trial.) Observation Word Equation Balanced Formula Equation Type Of Reaction F. Place A Small Apr 10th, 2024

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Name: Types Of Chemical Reactions Lab Introduction

Types Of Chemical Reactions Lab Introduction In This Unit, You Have Studied Various Types Of Chemical Reactions. Now You Will Transfer This Theoretical Knowledge Of Chemical Reactions To The Laboratory. In This Activity You Will Observe And Predict Products For Four Types Of Chemical Reactions Jan 15th, 2024

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